

Raman-Spectroscopic Observation of the Σ -diads of $^{12}\text{C}^{16}\text{O}^{18}\text{O}$ and $^{12}\text{C}^{16}\text{O}^{17}\text{O}$

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The Fermi resonance doublets of $^{12}\text{C}^{16}\text{O}^{18}\text{O}$ and $^{12}\text{C}^{16}\text{O}^{17}\text{O}$ were identified as weak Q-branches in the Raman spectrum of natural CO_2 .

Carbon dioxide and its isotopic variants have been the subject of many experimental (IR and Raman) and theoretical spectroscopic investigations^{2,3}. Surprisingly the Raman spectra of $^{12}\text{C}^{16}\text{O}_2$ and $^{13}\text{C}^{16}\text{O}_2$ only have been measured⁴ and only the former has been analysed for the vibration-rotational structure⁵.

In our program of experimental and theoretical Raman spectroscopic studies^{6–8} of gases we have also recorded rovibrational spectra of various isotopic species of CO_2 . Our experimental setup has been described previously^{6–8}. We also made use of a method for direct scanning of the isotropic part ($\sim 45^\circ$) of the scattered light⁹ in order to identify the Q-branches which may otherwise get submerged in the rotational wings.

In the isotropic part of the rovibrational spectrum of natural CO_2 in addition to several hitherto unobserved Q-branches of hot bands¹⁰ we identified four lines, the positions of which are given in Table 1, as to be the Σ -diads of $^{12}\text{C}^{16}\text{O}^{18}\text{O}$ and $^{12}\text{C}^{16}\text{O}^{17}\text{O}$. Their intensities are in agreement with the natural abundance of 0.4% $^{12}\text{C}^{16}\text{O}^{18}\text{O}$ and 0.07% $^{12}\text{C}^{16}\text{O}^{17}\text{O}$. The position of the lines was determined

Table 1. Observed and calculated wavenumbers of the Σ -diads of $^{12}\text{C}^{16}\text{O}^{18}\text{O}$ and $^{12}\text{C}^{16}\text{O}^{17}\text{O}$.

	exp.	calc.
$^{12}\text{C}^{16}\text{O}^{18}\text{O}$	1259.40 (± 0.05)	1259.42
	1365.88 (± 0.05)	1365.84
$^{12}\text{C}^{16}\text{O}^{17}\text{O}$	1272.29 (± 0.05)	1272.31
	1376.23 (± 0.2)	1376.05

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² Z. Cihla and A. Chedin, J. Mol. Spectrosc. **40**, 337 [1971] and references therein.

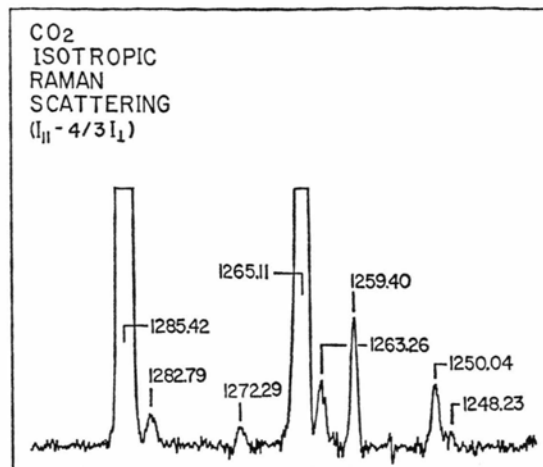


Fig. 1. Section of the purely isotropic part ($I_{\parallel} - 4/3 I_{\perp} \sim 45^\circ$) of the vibrational-rotational Raman spectrum of natural CO_2 . Gas pressure 1 atm., excitation 514.5 nm, 7 W. Slit width 1 cm^{-1} . Wavenumbers in cm^{-1} .

relative to the neighbouring rotational lines, the wavenumbers of which were calculated with rovibrational constants given by Dupré-Maquaire and Pinson³. Thus an accuracy of $\pm 0.05 \text{ cm}^{-1}$ could be obtained, with exception of the weakest line at $(1376.2 \pm 0.2) \text{ cm}^{-1}$. One section of the spectrum has been published earlier⁹, therefore we restrict ourselves to present in Fig. 1 another section showing the lower wavenumber components of the Σ -diads. The spectrum was obtained by subtracting $4/3$ of the $I_{\perp}$ component, which is proportional to $3\gamma'^2$, from the $I_{\parallel}$ component, which is proportional to $45\alpha'^2 + 4\gamma'^2$, thus resulting in a spectrum showing only the polarized Q-branches. It is interesting to note that both lines of $^{12}\text{C}^{16}\text{O}^{18}\text{O}$ are also discernible in the spectrum photographed by Stoicheff⁵, however, one was labelled as a ghost and the other one escaped attention.

Using the spectroscopic constants given by Jobard and Chedin¹¹ for $^{12}\text{C}^{16}\text{O}^{18}\text{O}$ and $^{12}\text{C}^{16}\text{O}^{17}\text{O}$ we calculated the Σ -diad wavenumbers¹². Table 1 shows that they are in excellent agreement with our observation. A comparison with a preliminary scan of enriched $^{12}\text{C}^{16}\text{O}^{18}\text{O}$ lends strong support to our assignments.

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- ¹¹ I. Jobard and A. Chedin, *J. Mol. Spectrosc.* **57**, 464 [1975].
- ¹² Although using Berney and Eggers¹³ constants one arrives at almost the same values for $^{12}\text{C}^{16}\text{O}^{18}\text{O}$, we preferred Jobard and Chedin's numbers, because their derivation overcomes earlier inconsistencies pointed out by Amat and Pimbert¹⁴.
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- ¹⁴ G. Amat and M. Pimbert, *J. Mol. Spectrosc.* **16**, 278 [1965].